Mat E 272

Lecture 19: Cast Irons

November 8, 2001

Introduction:

One reason steels and cast iron alloys find such wide-ranging applications and dominate industrial metal production is because of how they respond to thermal processing, or heat treatment. For a given alloy composition, one can obtain a rich and diverse range of microstructures simply by applying different combinations of time and temperature. We have seen how cooling rates affect microstructure; slowly cooled steels develop a coarse pearlitic microstructure while quenched steels undergo a martensitic (diffusionless) transformation. Intermediate cooling rates also produce varying amounts of bainite, in addition to pearlite or martensite. Austenite, the high temperature FCC polymorph of steel, can also be retained at room temperature by rapid cooling, albeit metastable. Spherodite is produced by heating pearlitic or bainitic microstructures at a temperature just below the eutectoid, or A1. All of these microstructures give rise to different mechanical properties, and it is incumbent on us to understand the relationships between composition, thermal history, and mechanical behavior. In addition, the characteristics of high-carbon Fe-C alloys, or cast irons, will be discussed.

Cast Irons

Definition:

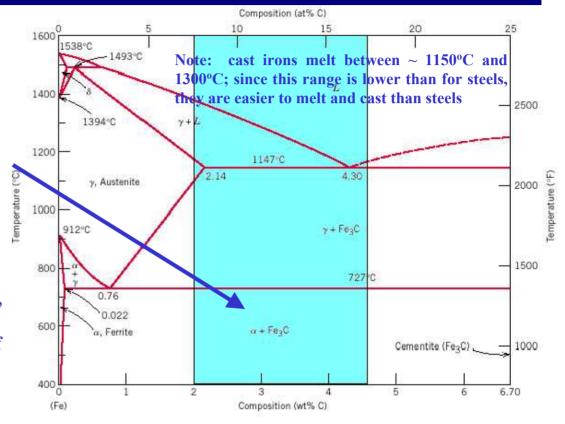
Cast iron is the name applied to a family of high-carbon content Fe-C alloys, specifically, those containing more than 2.14 wt. % C. Generally, most cast irons fall within the 3.0 to 4.2 wt. % range. Many contain silicon.

Applications:

Many cast irons are strong, but also brittle. As such, they find uses as small cylinder blocks, cylinder heads, pistons, clutch plates, transmission cases, diesel engine casting.. There are four types of cast irons: gray, nodular, white, and malleable.

Constituents:

Cast irons consists of varying amounts of ferrite, cementite (Fe₃C), and graphite (C). Note that for most cast irons, the carbon-rich phase is graphite, not cementite.



Fun fact: Did you know that iron is obtained by processing iron ore? Iron ore typically contains the following components:

- Hematite Fe2O3 70% iron
- •Magnetite Fe3O4 72% iron
- •Limonite Fe2O3 + H2O 50% to 66% iron
- •Siderite FeCO3 48% iron

Gray Cast Iron

Components:

Gray iron contains 2.5 to 4.0 wt. % C and also 1.0 to 3.0 wt. % silicon. The Si promotes formation of graphite instead of cementite.

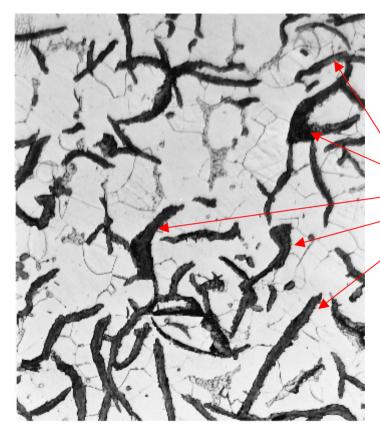
In contrast, sulfur promotes formation of carbide over graphite

Microstructure:

Graphite is dispersed throughout a ferrite or pearlite matrix in the form of *flakes*.

Characteristics:

The graphite flakes have sharp edges and tips. Consequently, the flakes act as *stress raisers* which can induce fracture near their tips. For this reason, gray iron is brittle in tension. It is also good at damping vibrations.—Gray iron is inexpensive and relatively easy to cast with minimal shrinkage.



Graphite flakes --

makes a fracture surface appear gray in color

→ Gray iron is used in support structures for large, heavy machines that generate significant vibration.

Nodular (Ductile) Iron

Components:

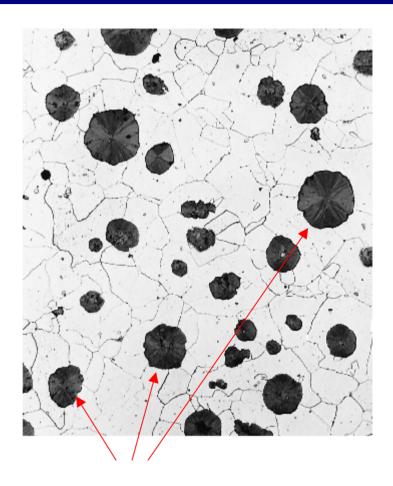
Nodular iron contains 2.5 to 4.0 wt. % C, 1.0 to 3.0 wt. % Si, plus Mg &/or Ce. The Mg &/or Ce cause the flakes to spherodize (hence nodular)

Microstructure:

Graphite is dispersed throughout a ferrite or pearlite matrix in the form of *spheres* or *nodules*.

Characteristics:

The graphite nodules have no sharp features; therefore, the resulting material is much more ductile than gray iron. The mechanical properties of nodular iron are similar to steel. Nodular iron is frequently used in valves, crankshafts, gears, and other automotive components.



Note the nodules of pure carbon (graphite) The lighter matrix is ferrite.

White Iron

Components:

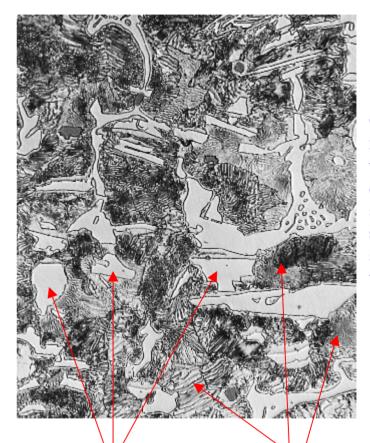
White iron contains 2.5 to 4.0 wt. % C and less than 1.0 wt. % Si. Because of the low Si content, the carbon forms Fe_3C instead of graphite.

Microstructure:

Cementite is dispersed throughout pearlite (ferrite + cementite) matrix.

Characteristics:

White iron contains a considerable volume fraction of cementite (Fe₃C), a hard and brittle compound. Because of the amount of cementite, white iron is extremely hard and extremely brittle. Limited usage - mainly applications requiring hardness and wear resistance (such a rollers in rolling mills). Also used as a precursor to malleable iron.



Cementite
is silvery white in
color; hence
a fracture
surface
appears
white

Lighter regions = Fe₃C; darker matrix = pearlite (ferrite + Fe₃C)

Malleable Iron

Components:

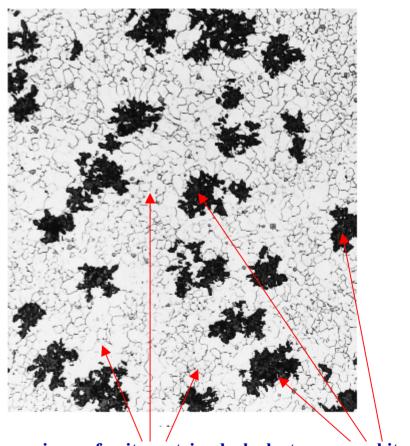
Malleable iron (just like white iron) contains 2.5 to 4.0 wt. % C and *less than* 1.0 wt. % Si. Unlike white iron, the C exists in the form of *graphite* instead of cementite.

Microstructure:

Graphite rosettes are dispersed throughout a *ferrite* (or pearlite) matrix.

Characteristics:

Malleable iron is produced by heating white iron in order to decompose the cementite into graphite. The graphite forms clusters, similar to nodular iron. Reduction in the amount of cementite causes the material to become relatively ductile (or malleable). Used in connecting rods, transmission gears, differential cases, flanges, pipe fittings.



Lighter regions = ferrite matrix; dark clusters = graphite

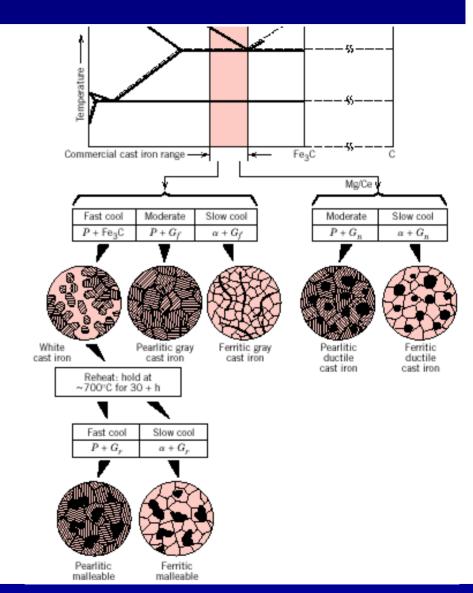
Cast Iron microstructures

An amazing diversity of microstructures can be obtain in cast iron simply by modifying the composition and adjusting the cooling rate.

In the schematic at right,

- Gf = flake graphite,
- Gr = graphite rosettes,
- Gn = graphite nodules,
- P = pearlite (which is α + Fe₃C), and
- α = ferrite.

Note that, in general, slow cooling tends to produce large microstructural features whereas rapid cooling results in fine (or small) features. Fine microstructures tend to be stronger but more brittle.



Cast Irons - Summary of Mechanical Properties

Table 12.5 Designations, Minimum Mechanical Properties, Approximate Compositions, and Typical Applications for Various Gray, Nodular, and Malleable Cast Irons

				Ме	chanical Prope	erties	
Grade	UNS Number	Composition (wt%) ^a	Matrix Structure	Tensile Strength [MPa (ksi)]	Yield Strength [MPa (ksi)]	Ductility [%EL in 50 mm (2 in.)]	Typical Applications
Gray Iron							
SAE G1800	F10004	3.40-3.7 C, 2.55 Si, 0.7 Mn	Ferrite + Pearlite	124 (18)	_	_	Miscellaneous soft iron castings in which strength is not a primary consideration
SAE G2500	F10005	3.2-3.5 C, 2.20 Si, 0.8 Mn	Ferrite + Pearlite	173 (25)	_	_	Small cylinder blocks, cylinder heads, pistons, clutch plates, transmission cases
SAE G4000	F10008	3.0-3.3 C, 2.0 Si, 0.8 Mn	Pearlite	276 (40)	_	_	Diesel engine castings, liners, cylinders, and pistons
Ductile (Nodular) Iron							
ASTM A536							
60-40-18	F32800	3.5-3.8 C, 2.0-2.8 Si,	Ferrite	414 (60)	276 (40)	18	Pressure-containing parts such as valve and pump bodies
100-70-03	F34800	0.05 Mg, <0.20 Ni,	Pearlite	689 (100)	483 (70)	3	High-strength gears and machine components
120-90-02	F36200	<0.10 Mo	Tempered martensite	827 (120)	621 (90)	2	Pinions, gears, rollers, slides
Malleable Iron							
32510	F22200	2.3-2.7 C, 1.0-1.75 Si, <0.55 Mn	Ferrite	345 (50)	224 (32)	10	General engineering service at normal and ele- vated temperatures
45006	_	2.4-2.7 C, 1.25-1.55 Si, <0.55 Mn	Ferrite + Pearlite	448 (65)	310 (45)	6)	

Gray vs. Nodular Irons - a comparison

Gray irons --

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a composite of graphite + high carbon iron
extremely brittle
stress concentration at the tips of graphite flakes
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Nodular irons --

amazing change in mechanical properties accompanies a change in microstructure graphite spheres (nodules) replace flakes

tensile strength can be as high as 100,000 psi with favorable ductility

Typical properties:

ductility: 2 - 30% elongation in 2 inches

tensile: 55,000 to 120,000 psi (375 - 818 MPa)

yield: 30,000 to 90,000 psi (205 - 614 MPa)